

Who

4

S

mater

by Ellen Lawrence

Salfu?

[Intentionally Left Blank]



by Ellen Lawrence

Consultant:

Shawn W. Wallace Department of Earth and Planetary Sciences American Museum of Natural History New York, New York



New York, New York

Credits

Cover, © Vibrant Image Studio/Shutterstock; 4, © Dmitry Naumov/Shutterstock; 5T, © NOAA/NASA; 5, © Stanislav Smoliakov/ Shutterstock; 6T, © M. Unal Ozmen/Shutterstock; 6B, © Channonttennis/Shutterstock; 7, © jordeangjelovik/Shutterstock; 8, © Szabi/Shutterstock; 9, © CnOPhoto/Shutterstock; 10, © Rafa Irusta/Shutterstock; 11, © Shutterstock; 12, © NOAA PMEL Vents Program/Science Photo Library; 13, © EdStock/iStock; 14, © Ruby Tuesday Books; 15, © Blue Ice/Shutterstock; 16T, © Cosmographics; 16B, © kavram/Shutterstock; 17, © ChameleonsEye/Shutterstock; 18, © KPG Payless2/Shutterstock; 19, © bikeriderlondon/Shutterstock; 20, © tisskananat/Shutterstock; 21T, © Gayvoronskaya Yana/Shutterstock; 21, © stable/Shutterstock; 22, © M. Unal Ozmen/ Shutterstock, © Ann Worthy/Shutterstock, and © Jiang Hongyan/Shutterstock; 23TL, © Lucky-photographer/Shutterstock; 23TC, © beboy/Shutterstock; 23TR, © Yuriy Kulik/Shutterstock; 23BL, © isarescheewin/Shutterstock; 23BC, © Sayan Puangkham/ Shutterstock; 23BR, © Vilainecrevette/Shutterstock.

Publisher: Kenn Goin Editor: Jessica Rudolph Creative Director: Spencer Brinker Design: Emma Randall Photo Researcher: Ruby Tuesday Books Ltd

Library of Congress Cataloging-in-Publication Data

Names: Lawrence, Ellen, 1967– author. Title: Why is seawater salty? / by Ellen Lawrence. Description: New York, New York : Bearport Publishing, [2016] | Series: Drip, drip, drop: Earth's water | Audience: Ages 6–10. | Includes bibliographical references and index. Identifiers: LCCN 2015040326 (print) | LCCN 2015041634 (ebook) | ISBN 9781943553228 (library binding) | ISBN 9781943553563 (ebook) Subjects: LCSH: Seawater—Composition—Juvenile literature. | Salt—Juvenile literature. | Salt industry and trade—Juvenile literature. Classification: LCC GCI21.2 .L39 2016 (print) | LCC GCI21.2 (ebook) | DDC 551.46/5—dc23 LC record available at http://lccn.loc.gov/2015040326

Copyright © 2016 Bearport Publishing Company, Inc. All rights reserved. No part of this publication may be reproduced in whole or in part, stored in any retrieval system, or transmitted in any form or by any means, electronic, mechanical, photocopying, recording, or otherwise, without written permission from the publisher.

For more information, write to Bearport Publishing Company, Inc., 45 West 21st Street, Suite 3B, New York, New York 10010. Printed in the United States of America.

10987654321

Contents

What Makes the Ocean Salty?6 💧	
Rocks and Salt8	
Saltier and Saltier 10	
Under the Sea I2	
Saltwater Lakes 14	
The Dead Sea 16 🦱	
Can We Drink Salt Water? 18	
Collecting Salt 20	
Science Lab	
Science Words 23	
Index	
Read More 24	
Learn More Online	
About the Author	

Salty Seawater

It's a sunny day at the beach, and you're swimming in the ocean.

You accidentally swallow a mouthful of salty seawater. Yuck!

The water that comes out of a faucet doesn't taste salty.

If you catch raindrops on your tongue, they don't taste like salt.

So what makes seawater so salty?







What Makes the Ocean Salty?

Seawater is salty because it contains tiny **particles** of salt.

In fact, there's nearly I teaspoon (5 ml) of salt in I pint (473 ml) of seawater.

The particles of salt are too tiny to see, but you can taste them. 450 m/ 16 07 400 ml 14 07 350 ml 12 oz 4LITRE 1/2 PINT 200 ml 8 oz 6 oz a teaspoon of salt If all the salt in the oceans was removed and spread evenly over all the land on Earth, how deep would it be? The layer of salt would be 500 feet (152 m) deep. That's the height of a 40-story building!

How do you think salt gets into the oceans?